

Docket No. 220053US0X PCT

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

IN RE APPLICATION OF: Kiyoshi FUDA, et al.

SERIAL NO: 10/070,756

GAU: 1764

FILED: June 26, 2002

EXAMINER:

FOR: METHOD OF TREATING FLUORINE COMPOUND

RECEIVED
JUL 28 2003
GROUP 1700

INFORMATION DISCLOSURE STATEMENT UNDER 37 CFR 1.97

COMMISSIONER FOR PATENTS
ALEXANDRIA, VIRGINIA 22313

SIR:

Applicant(s) wish to disclose the following information.

REFERENCES

- ☐ The applicant(s) wish to make of record the references listed on the attached form PTO-1449. Copies of the listed references are attached, where required, as are either statements of relevancy or any readily available English translations of pertinent portions of any non-English language references.
- ☐ A check is attached in the amount required under 37 CFR §1.17(p).

RELATED CASES

- ☒ Attached is a list of applicant's pending application(s) or issued patent(s) which may be related to the present application. A copy of the claims and drawings of the pending application(s) is attached.
- ☐ A check is attached in the amount required under 37 CFR §1.17(p).

CERTIFICATION

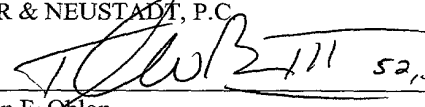
- ☐ Each item of information contained in this information disclosure statement was first cited in a communication from a foreign patent office in a counterpart foreign application not more than three months prior to the filing of this statement.
- ☐ No item of information contained in this information disclosure statement was cited in a communication from a foreign patent office in a counterpart foreign application or, to the knowledge of the undersigned, having made reasonable inquiry, was known to any individual designated in 37 CFR §1.56(c) more than three months prior to the filing of this statement.

DEPOSIT ACCOUNT

- ☒ Please charge any additional fees for the papers being filed herewith and for which no check is enclosed herewith, or credit any overpayment to deposit account number 15-0030. A duplicate copy of this sheet is enclosed.

Respectfully submitted,

OBLON, SPIVAK, McCLELLAND,
MAIER & NEUSTADT, P.C.


Norman E. Oblon

Registration No. 24,618



22850

Tel. (703) 413-3000
Fax. (703) 413-2220
(OSMMN 05/03)



LIST OF RELATED CASES

RECEIVED
JUL 28 2003
GROUP 1700

<u>Docket Number</u>	<u>Serial or Patent Number</u>	<u>Filing or Issue Date</u>	<u>Inventor/ Applicant</u>
220053US0X PCT*	10/070,756	06/26/02	FUDA, et al.
220036US0X PCT	10/088,073	06/12/02	FUDA, et al.

*Present Application; listed for information
NFO/sbr

220036

What is claimed is.

[1] A treatment process of a solution containing an organic compound having a fluorocarbon chain (hereinafter said to as the fluorine compound), the process comprising,
adding divalent and trivalent metal salts to said solution,
forming a layered double hydroxide having the fluorine compound between layers to absorb and fix the fluorine compound.

[2] A treatment process of a solution containing the fluorine compound, the process comprising,
adding divalent and trivalent metal salts to said solution,
precipitating a layered double hydroxide having the fluorine compound between layers,
separating a solid part by the solid-liquid separation,
dissolving said separated solid part in an acid, and
separating the fluorine compound or its salt.

[3] The treatment process of the solution containing the fluorine compound according to claims [1] or [2], the process further comprising,
adjusting pH of the solution to more than 4,
precipitating the layered double hydroxide having the fluorine compound between layers.

[4] The treatment process of the solution containing the fluorine compound according to claims [1] or [2], the process further comprising,
adding an alkali to the solution to adjust pH from 4 to 12,
adding divalent and trivalent metal salts to said solution
precipitating the layered double hydroxide having the fluorine compound

FOR INFORMATION
DISCLOSURE
PURPOSES ONLY

Related Pending Application
Related Case Serial No: 10/088,073
Related Case Filing Date: 10-12-02

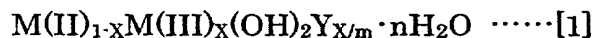
between layers.

[5] The treatment process of the solution containing the fluorine compound according to any one of claims [1] to [4], wherein the divalent metal salt is a salt of magnesium, calcium, zinc, nickel, copper, manganese (divalent), or cobalt (divalent), and the trivalent metal salt is a salt of aluminum, iron, chromium, manganese (trivalent), cobalt (trivalent), potassium, lanthanum, or scandium.

[6] The treatment process of the solution containing the fluorine compound according to any one of claims [1] to [5], wherein the divalent and the trivalent metal salts are chlorides.

[7] The treatment process of the solution containing the fluorine compound according to any one of claims [1] to [6], wherein the fluorine compound is carboxylic acid or sulfonic acid having the fluorocarbon chain, in which the number of carbon is more than 5.

[8] The treatment process of the solution containing the fluorine compound according to any one of claims [1] to [7], wherein the layered double hydroxide having the fluorine compound between layers is shown in the following formula [1].



where, Y is an anion having valence number m of the fluorine compound having the fluorocarbon chain, M(II) is a divalent metal ion, M(III) is a trivalent metal ion, X is 0.1 to 0.5, and n is 0 or positive integer.

[9] A treatment process for recovering the fluorine compound and its salts, the process comprising, precipitating the layered double hydroxide by the treatment process

according to any one of claims [1] to [7],
recovering the solid part by the solid-liquid separation,
dissolving said recovered solid part in a mineral acid to recover the
separated fluorine compound or its salts, or
heating said mineral acid dissolving the recovered solid part,
putting quietly to separate an oil layer, and
taking out the oil layer to recover the fluorine compound and its salts.

[10] A treatment process for recovering a fluorine compound and its
salts, the process comprising,
precipitating the layered double hydroxide by the treatment process
according to any one of claims [1] to [7],
recovering the solid part by the solid-liquid separation,
dispersing the recovered solid part to an organic solvent, and
filtering an insoluble part from said solvent.

[11] A layered double hydroxide shown by above-mentioned formula
[1], which contains the fluorine compound between layers and is formed by
adding the divalent and trivalent metal salts to the solution containing the
fluorine compound having the fluorocarbon chain.

Summary

Divalent and trivalent metal salts are added to the solution containing the fluorine compound to precipitate the layered double hydroxide containing the fluorine compound between layers. By these processes, the fluorine compound can be fixed with high rate. Moreover, if necessary, the precipitated layered double hydroxide can be recovered to separate the fluorine compound or its salt between layers. Therefore, the burden to environment or the ecosystem by the fluorine compound can be reduced.

Table 1

No.	Fluorine Compound Solution			Additional Metal Salt		Solution	Precipitate		
	Kinds	Concentration, Amount	Amount of Anion	Divalent	Trivalent		Amount	Anion Concentration	Anion Fixing Ratio
1	Per-fluoro-octanoic acid ammonium ($C_7F_{15}COONH_4$)	Aqueous solution of 0.1%, 1000ml	2.32mmol	Zn 4.64mmol	Al 2.32mmol	7	1.50 g	2.23mmol	96.1%
2	Ditto	Aqueous solution of 0.1%, 100ml Aqueous solution of 0.01%, 1000ml	2.32mmol 0.232mmol	Zn: 4.64mmol Zn: 0.464mmol	Al: 2.32mmol Al: 0.232mmol	7	1.50 g 0.15 g	2.3mmol 0.21mmol	99.1% 91.0%
3	Ditto	Aqueous solution of 0.1%, 1000ml	2.32mmol	Zn 4.64mmol	Al 2.32mmol	5	1.50 g	2.27mmol	98%
4	Ditto	Aqueous solution of 0.1%, 1000ml	2.32mmol	Mg 4.64mmol	Al 2.32mmol	10	1.10 g	1.4mmol	80%
5	Ditto	Aqueous solution of 0.1%, 1000ml	2.32mmol	Ca 4.64mmol	Al 2.32mmol	10	0.85 g	2.3mmol	60%
6	Per-fluoro-octyl-sulfonic acid ammonium ($C_8F_{17}SO_3NH_4$)	Aqueous solution of 0.1%, 1200ml	2.32mmol	Zn 4.64mmol	Al 2.32mmol	7	1.70 g	2.23mmol	99%
7	Per-fluoro-octyl-sulfonic acid lithium ($C_8F_{17}SO_3Li$)	Aqueous solution of 0.1%, 1174ml	2.32mmol	Zn 4.64mmol	Al 2.32mmol	7	1.65 g	2.21mmol	95%
10	Per-fluoro-decanoic acid ammonium ($C_9F_{19}COONH_4$)	Aqueous solution of 0.005%, 1000ml	1.16mmol	Zn 2.32mmol	Al 1.16mmol	7	0.84 g	1.09mmol	94%